



Karnataka State Council for Science and Technology

Indian Institute of Science Campus, Bengaluru - 560 012

Telephone: 080-23341652, 23348848, 23348849 ♦ Telefax: 080-23348840

Email: office@kscst.iisc.ernet.in, office@kscst.org.in ♦ Website: www.kscst.iisc.ernet.in, www.kscst.org.in

Mr. H. Hemanth Kumar
Executive Secretary

29
27th March 2019

Ref: 7.1.01/SPP/1333

The Principal,
Sapthagiri College of Engineering,
#14/5, Chickasandra,
Hesaraghatta Main Road,
Bengaluru - 560 057.

Dear Sir/Madam,

Sub : Sanction of Student Project (Biofuel) - 42nd Series: Year 2018-2019

Your Project Proposal Reference No. : **42S_B_BE_003**

Ref : Your Project Proposal entitled " **BIOPLASTIC EXTRACTION FROM WASTE GREASE PRODUCED IN INDUSTRIES USING AS GLYCEROL AS A SUBSTRATE**

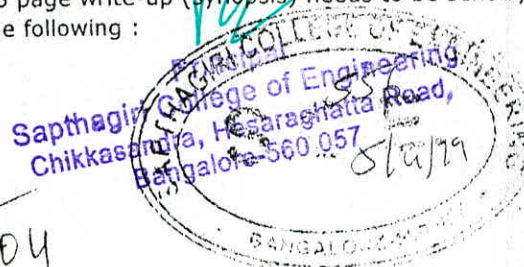
I am happy to inform that your student project proposal referred above, has been approved by the Council for "Student Project Programme (Biofuel)- 42nd Series" and has been sanctioned with a budgetary break-up as detailed below:

Student / s	Ms. Debika Chakrabarty and others	Budget	Amount (Rs)
		Materials/Consumables	3,500.00
Guide/s	Mr. Prashanth Kumar H P Ms. Shobha G	Labor	-
		Travel	500.00
Department	Biotechnology	Analysis	500.00
		Miscellaneous	500.00
		Report	500.00
		TOTAL	5,500.00
RUPEES FIVE THOUSAND FIVE HUNDRED ONLY			

The following are the guidelines to carryout the project work :

- The project should be performed based on the objectives of the proposal sent by you.
- The project should be completed in all respects and one copy of the hardbound report along with softcopy of the full report in a CD (.pdf format) should be submitted to KSCST.
- The project report shall mention the name of "Karnataka State Bioenergy Development Board and Karnataka State Council for Science and Technology" as sponsored organisations in the title page. Project Title or the objectives can be altered only with prior permission of KSCST. Any change in the project are strictly prohibited and liable for rejection and the amount sanctioned has to be returned back to KSCST. The fund is to be utilised only for the activities to which it has been released.
- Please quote your project reference number printed above in all your future correspondences.
- Important:** After completing the project, 2 to 3 page write-up (synopsis) needs to be sent by e-mail [biofuelcell.kscst@gmail.com] and should include following :

- Title of the project
- Name of the College & Department
- Name of the students & Guide(s)
- Keywords



Payment Voucher

No. 1073

Dated 7-Sep-2019

Particulars	Amount
Account : KSCST(Project Work)	5,500.00

Through :

SBI- 30263521464

On Account of :

Being funds released towards KSCST Project
work on Bio Plastic Extractin of waste Grease
using Glyceral as substrate


Bank Transaction Details:

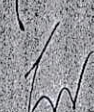
Debika Chakraborty
Cheque 499736 7-Sep-2019 5,500.00

Amount (in words) :

INR Five Thousand Five Hundred Only

₹ 5,500.00


R. S. AKASH (1591085023)
Receiver's Signature


Authorised Signatory




Principal
Sapthagiri College of Engineering
Chikkasandra, Hesaraghatta Road,
Bangalore-560 057

From,

14/08/19

Bangalore

Debika.C 7019226230

Department of Biotechnology

Sapthagiri College of Engineering

Bangalore - 560057

To,

The Principal

Sapthagiri College of Engineering

Bangalore - 560057

Respected Sir,

Subject: To release the fund
sanctioned from KSCST

I, Debika C and one groupmate Dishu D B, Kceethna N, B Vaishini have successfully completed the project on "Bioplastic extraction from waste grease using glycerol as substrate". We have submitted the report and CD to SPP Coordinator Ravishankar M N. I hereby request you to release the sanctioned amount for the project.

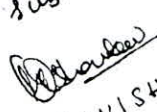
Thanking You,

Yours faithfully,

Debika.C

Principal
Sapthagiri College of Engineering
Chikkasandra, Hesaraghatta Road,
Bangalore-560 057


The above project was funded by KSCST
The students have completed the project
and submitted the report.


RAVISHANKAR M N
KSCST-SPP Coordinator
SCE Block
14/08/19

14/8/19

DETAILS ARE FURNISHED AS BELOW:

Sl. No	Items	Cost (in Rupees)
1	Raw materials	300
2	Chemicals	1000
3	Glass wares	250
4	Demand draft	2000
5	Project Report	1200
6	Miscellaneous (Transportation, Xerox, writing materials etc.,)	750
	Total Sum	5500 ✓
<u>Amount in words</u> : Five thousand five hundred only		


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Bangalore-560 057


14/08/19

Mob : 9945754390

MAHESH CYBER WORLD

103/2, Rukmini Nagar, Nagasandra (P), Bangalore - 73

No.

Date 24/11/19

M/s

Deleka C

No.	Particulars	Qty.	Amount
1	INTERNET	1	100
2	PRINT OUT		800
3	XEROX		100
4	DTP		500
5	COMPUTER SERVICE		
6	RENT AGREEMENT		
	TOTAL		1500/-

MAHESH CYBER WORLD
Sapthagiri Engg College
Chikkasandra, Bangalore-560057

Kov
Principal
Sapthagiri College of Engineering
Chikkasandra, Hesaraghatta Road,
Bangalore-560 057

B
Principal
Sapthagiri College of Engineering
Chikkasandra, Hesaraghatta Road,
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(ORIGINAL FOR RECIPIENT)



Invoice No. 0535	Dated 17-Jul-2019
Delivery Note	Mode/Terms of Payment Immediate
Supplier's Ref. 0535	Other Reference(s)
Buyer's Order No. Verble	Dated 17-Jul-2019
Despatch Document No.	Delivery Note Date
Despatched through By Hand	Destination
Terms of Delivery	

Miss, Keerthana (Sapthagirl College of Engineering)
No 14/5, Chikkasandra, Heasaragatta Main
Road, Bangalorre-560057
State Name : Karnataka, Code : 29

Principal
Sapthagiri College of Engineering
Bikkasandra, Hesaraghatta Road
Bangalore-560 057

42S_B_BE_003 Bioplastic Extraction from waste grease produced in industries using as glycerol as a substrate

COLLEGE:Sapthagiri College Of Engineering, Bangalore

DEPARTMENT:Biotechnology

GUIDE:Mr Prashanth Kumar and Mrs Shobha G

STUDENTS:Debika Chakrabarty, Disha DB, Keerthana N, B Varshini

INTRODUCTION

The used cooking oil and trap grease can contain much more than 15% FFAs. These feedstock need additional processing before they can undergo traditional alkali-catalysed transesterification to form glycerol and Biodiesel.

This process converts triacylglycerol and methanol into glycerol and fatty acid methyl esters (namely biodiesel) using alkali or acid catalysts. The amount of crude glycerol produced from this transesterification reaction accounts for approximately 10% of the final weight of biodiesel. Conversion of crude glycerol into higher-value products improves the economic viability of biofuel industry by coupling the production of value-added products to the production of biodiesel and eliminating the cost of treatment for crude glycerol disposal. Fermentation of glycerol has been reported to produce many value-added by-products, such as 1,3-propanediol, dihydroxyacetone, succinic acid, propionic acid, ethanol, butanol, hydrogen, citric acid, lactic acid, glyceric acid, bio surfactants, pigments, and PHAs. Among these by-products, 1,3-propanediol, 75 succinic acid, lactic acid, and glyceric acid, have been used as bio monomers for production of plastics, i.e. polyesters, polyethers, and polyurethanes, through chemical synthesis. In the present work glycerol obtained after transesterification will be used for bio plastic formation.

Materials and methodology:

Production of Glycerol:

Transesterification of waste grease using alkaline catalyst was carried out using a two-step process. The two-step reaction utilized 100% excess methanol, 6:1 molar ratio of alcohol to oil and 1% KOH as a catalyst. In each step, 3 mol of alcohol and 0.5% KOH were used and reaction was carried out at 25 °C

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for 30 min. After the first step the waste grease having a high free fatty acid formed a thick soap which interfered with the glycerol separation (Issariyakul *et al.*, 2006).

Test for Glycerol:

Acrolein test

Acrolein test is used to detect the presence of glycerol or fat. When fat is treated strongly in the presence of a dehydrating agent like potassium bisulphate (KHSO_4), the glycerol portion of the molecule is dehydrated to form an unsaturated aldehyde, acrolein that has a pungent irritating odour.

Dichromate test:

In a dry test tube 3 or 4 ml of glycerol solution, to it add a few drops of 5% potassium dichromate solution and 5 ml of conc. HNO_3 , mix well and note that the brown colour is changed to blue.

Dunstan's test :

A drop of phenolphthalein is added to approximately 5 mL of borax solution . The pink color appears. On adding 2-3 drops of glycerol, the pink color disappears on heating and disappears on cooling again.

Bioplastic production from grease:

Preparation of bioplastic was done by modifying the agar and the glycerol concentration.

1.5%, 3%, 4.5% of bioplastic was prepared. According to the concentration required amount of glycerol.

Moisture Absorption Test:

The moisture absorption test identified the ability of bioplastics to absorb water (H_2O) as determined by standard ASTM D 570. Bioplastics, which had been previously dried for 24 hours in an oven at 50°C , cooled in a desiccator, and weighed, were cut into 2mm x 2mm. The moisture absorption data of bioplastics was obtained by soaking them in water for 24 hours. After that, the bioplastics were dried with a cloth and immediately weighed.

$$\text{Moisture Content (\%)} = \frac{(\text{Post-BrakeWeight}) - (\text{InitialWeight})}{\text{InitialWeight}} \times 100$$

InitialWeight

Biodegradability

Biodegradable behaviour of bioplastics was determined using soil burial degradation test, i.e. bioplastics were buried in the soil, so that it would be degraded completely.[Degradation testing serves to determine the extent of damage of bioplastics. The damage can be seen from the mass reduction of respective specimens buried in the ground. Bioplastics were cut into 10 mm x 10 mm. Then, they were buried into the ground at 8-cm depth; the burial duration varied (3, 6, 9, and 12 days). Prior to burial, the initial mass (mass before degradation) was determined. The final mass (mass after degradation) of the bioplastics was measured afterwards. Any changes in mechanical properties due to degradation process were observed and when the bioplastics were completely degraded, the biodegradability was measured.

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$$\text{Microbial Resistance (\%)} = \frac{\text{Final Mass} - \text{Initial Mass}}{\text{Initial Mass}} \times 100$$

Results

Production of Glycerol:

Transesterification of waste grease was carried out and glycerol was obtained.

Test for Glycerol:

Acrolein test

Acrolein test is used to detect the presence of glycerol or fat. When fat is treated strongly in the presence of a dehydrating agent like potassium bisulphate (KHSO_4), the glycerol portion of the molecule is dehydrated to form an unsaturated aldehyde, acrolein that has a pungent irritating odour was observed.

Dichromate test:

In a dry test tube 3 or 4 ml of glycerol solution, to it a few drops of 5% potassium dichromate solution was added and 5 ml of conc. HNO_3 , was mixed well and there was a colour change from brown colour is changed to blue.

Dunstan's test :

A drop of phenolphthalein was added to approximately 5 ml of borax solution . The pink colour appears. On adding 2-3 drops of glycerol, disappearance of pink colour on addition of glycerol.

Bioplastic production from grease:

Preparation of bioplastic was done by modifying the agar and the glycerol concentration.

1.5%, 3%, 4.5% of bioplastic was prepared. According to the concentration required amount of glycerol.

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$$\text{Moisture Content (\%)} = \frac{(\text{Post-Brake Weight}) - (\text{Initial Weight})}{\text{Initial Weight}} \times 100$$

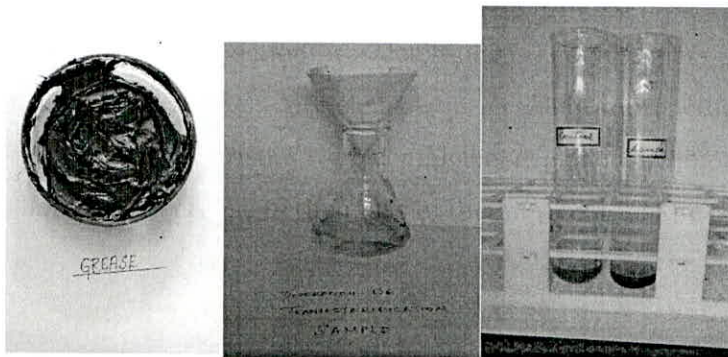
72.15% was moisture absorption was observed.

Biodegradability

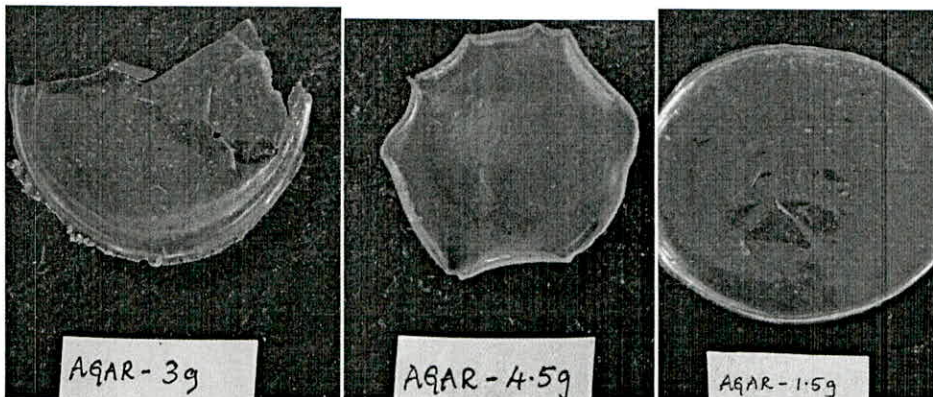
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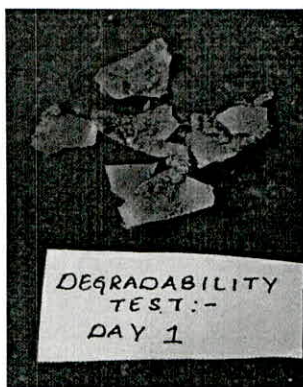
the extent of damage of bioplastics. The damage can be seen from the mass reduction of respective specimens buried in the ground. Bioplastics were cut into 10 mm x 10 mm. Then, they were buried into the ground at 8-cm depth; the burial duration varied (3, 6, 9, and 12 days). Prior to burial, the initial mass (mass before degradation) was determined. The final mass (mass after degradation) of the bioplastics was measured afterwards. Any changes in mechanical properties due to degradation process were observed and when the bioplastics were completely degraded, the biodegradability was measured. There was decrease in the mass of the bioplastic as the time was increased.



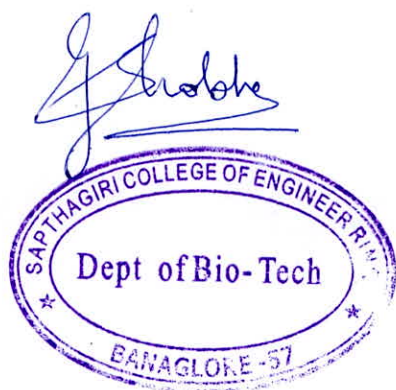
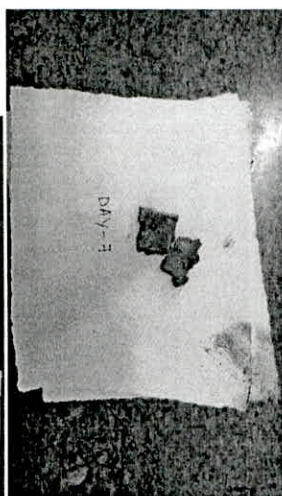
Dunstans test




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DEGRADABILITY
TEST:-
DAY 1




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